UTILIZATION OF CAPTURED CO₂ FOR IMPLEMENTING CCUS IN ROMANIA

Iolanda ALEXE¹, Mihai CHIRAN¹, Constantin Ştefan SAVA², Alexandra DUDU², Sorin ANGHEL², Andreea BURLACU³, Corneliu DINU³, Romeo SOARE³

¹Picoil Info Consult, 21 Ion Câmpineanu Street, Câmpina, Prahova, Romania, e-mail: iolanda.alexe@pic-oil.ro, mchiran@pic-oil.ro
²National Institute of Marine Geology and Geo-Ecology (GeoEcoMar), 23-25 Dimitrie Onciul St., 024053 Bucharest, Romania e-mail: savac@geoecomar.ro, alexandra.dudu@geoecomar.ro, soanghel@geoecomar.ro
³CO₂ Club, 23-25 Dimitrie Onciul St., 024053 Bucharest, Romania, e-mail: andreea@co2clubromania.com, codinu@gmail.com, romea.soare@aefdj.ro

Abstract. The promotion of the Carbon Capture and Utilisation (CCU) technology relies on the priorities of the European Commission that includes the Circular Economy as a major challenge (European Commission, 2018). To this end, under the EU Research and Innovation Programme (Horizon 2020), the Commission will demonstrate the opportunities for moving towards a circular economy at European level with large-scale innovation projects. Romania is part of this program, and is trying to implement CCU technologies in near future. While Carbon Capture and Storage (CCS) technologies are well known, studied and are starting to be implemented, the usage of captured CO₂ is less taken into account at present. These technologies involve the participation of more entities in several industrial branches, and the result is often only capturing CO₂ in a cycle, which eventually is reaching to the atmosphere but later and after being used in some technological processes.

Key words: captured CO₂ utilization, CO₂ utilization options, CO₂ chemical conversion, CO₂ biological conversion transformations of CO₂

1. INTRODUCTION

Carbon capture and storage (CCS) alone is a technology directed to CO₂ abatement by storing CO₂ captured from stationary industrial installations in suitable geological sites and preventing its release into the atmosphere.

CCU technologies instead are directed towards CO₂ abatement and also to converting captured CO₂ into useful products which can be valued and sold. Compared to CCS alone, CCU has already a business case. The CO₂ supply is guaranteed by various industries that will continue to emit CO₂, other than the energy industry (cement, steel, chemicals, glass etc.). One major problem of CCU could be that conversion of CO₂ to various products is energy intensive due to thermodynamic stability of CO₂ (Cuéllar-Franca, Azapagic, 2015).

However, CCU alone cannot realistically fix all emissions due to the large volumes involved and the potentially relatively low markets for individual products. Global CCS Institute (2011) concluded that only a few per cent of total emissions could be utilized. Still, CCU can be a reasonable solution in countries or regions where CO₂ geological storage is not allowed or is confronted with a lack of public acceptance, as well as in countries with no geological storage capacity.

Different CCS and CCU options are summarized in Fig. 1.

The present paper intends to make a comprehensive review of different CCU technologies and options that will form the basis for evaluating CCU options that could be suitable for Romania. In Romania there is already a market, although relatively low, for CO₂ and the major emissions of CO₂ (more than 100 kt CO₂ per year) from stationary industrial installations have reached a total of more than 37 Mt of CO₂ (ANPM, 2018).
2. DIRECT UTILIZATION OF CAPTURED CO$_2$

CCU applications are diverse and associated with many industries. Figure 2 presents the utilisation options for captured CO$_2$ and their association with specific industries.

Talking about CCU options, the two forms of CO$_2$ utilisation are usually considered, directly or through conversion. Examples of direct utilisation of CO$_2$ include its use in the food and drink industry, in the pharmaceutical industry, the horticulture and for enhanced recovery of hydrocarbons. The first three utilisations require a high purity CO$_2$ stream which can be derived for example from ammonia production (Cuéllar-Franca, Azapagic, 2015).

In the food and drink industry, CO$_2$ is commonly used as a carbonating agent, (ex: carbonation of beverages with high-purity CO$_2$), as a seal gas to prevent oxidation of the wine during maturation, as a solvent for the extraction of flavours, extracting fat from food and in the decaffeination process (Cuéllar-Franca, Azapagic, 2015). In the pharmaceutical industry, CO$_2$ can be used as a respiratory stimulant, as an intermediate in the synthesis of drugs, or in surgery, sterilization and blood tests. CO$_2$ is used in horticulture to maintain optimal CO$_2$ concentration and maximise plant growth rate. Carbon dioxide systems dramatically improve the growth and quality of greenhouse plants. Increased gas concentrations lead to larger, healthier and faster growing plants and lower operating costs, especially during winter, when heating costs can be reduced by up to 50%. Carbon dioxide replaces gas generators, saving fuel costs and eliminating harmful emissions.

One of the best known uses of CO$_2$ is in fire extinguishers proving itself very efficient to combat fires when water is inefficient, unwanted or unavailable.

Enhanced oil and coal-bed methane recovery (EOR and ECBM) are other examples of direct utilisation of CO$_2$ where it is used to extract oil from an oil field or natural gas from unmineable coal deposits, respectively (Cuéllar-Franca, Azapagic, 2015). This type of CO$_2$ utilisation can lead to permanent storage of CO$_2$ (Bennett et al., 2014). Enhanced oil recovery using CO$_2$ injection or CO$_2$ flooding has been widely practiced for over 40 years (IEA, 2015) in several oil-producing countries including Norway, Canada and the USA. It is a mature technology which now focuses also to enhancing CO$_2$ storage, apart from recovery of hydrocarbons.

3. CHEMICAL CONVERSION AND UTILIZATION OF CAPTURED CO$_2$

Utilization of CO$_2$ through conversion processes is known and applied since the second half of nineteen century (Aresta et al., 2013) comprising mature technologies such as synthe-
sis of salicylic acid and urea production. New trends include mineral carbonation and production of biofuels using algae.

3.1. Conversion of CO₂ into chemicals and fuels

The conversion of CO₂ into chemicals and fuels can be accomplished by carboxylation or reduction reactions, leading to production of methane, methanol, syngas, urea and formic acid (Cuéllar-Franca, Azapagic, 2015). This type of conversion is very energy intensive and requires highly selective catalysts due to the low reactivity of CO₂.

3.2. Mineral carbonation

The concept of storage of CO₂ as calcium and magnesium carbonate minerals is commonly referred to as mineral carbonation (IPCC, 2005). In mineral carbonation, (captured) CO₂ is used with minerals (mostly calcium or magnesium silicates) to form Ca or Mg carbonates. As mineral feedstock, natural rocks that are rich in alkaline earth silicates can be used, such as woonlastonite, olivine and serpentine which are available in large quantities in many places around the world and could lead to permanent storage of billions of tonnes of CO₂ (Bennett et al., 2014). Mineral carbonation could be an alternative for long term geological storage, especially for regions where CO₂ underground storage is not possible. The main advantage of mineral carbonation is the formation of stable carbonates capable of storing CO₂ for long periods (decades to centuries), without the risk of CO₂ leakage as in CCS. Apart from natural minerals, for mineral carbonation, waste materials from cement and steel industry can be used providing also a way to recycle these products (Bennett et al., 2014). There is also a drawback of this technology. It will require major dumps of the newly formed mineral masses to be built which uses space and has significant costs.

3.3. CO₂ absorption by microalgae to generate biomass

CO₂ can be used to cultivate microalgae for the production of bio-oils and proteins, biofuels, chemicals, ingredients for food or cosmetics, soil conditioners or animal feed. Microalgae are microscopic, single-celled plants growing in fresh water or seawater. They use sunlight as their energy source, and CO₂ and inorganic ingredients, mainly N-compounds (NO₃⁻, NH₄⁺) and phosphates, for growth. They have the ability to fix CO₂ directly from waste streams such as flue gas as well as using nitrogen from the gas as a nutrient. Cultivation of microalgae can be carried out in open ponds and photo-bioreactors (Aresta et al., 2013; Cuéllar-Franca, Azapagic, 2015). The CO₂ sequestering process consists in “bubbling CO₂ through algal cultivation systems” (Global CCS Institute, 2011) which together with sunlight and water increases productivity / biomass generation in the algae cultivation process.

The first step in transforming microalgae into fuels is harvesting and drying them. Their transformation can be accomplished by thermochemical or biochemical conversions.
Thermochemical conversion consists of processes such as gasification, liquefaction and pyrolysis, processes in which heat and electricity are used. Biochemical conversion is based on biological and chemical processes, such as anaerobic digestion, fermentation and esterification.

The following diagram, Figure 3, shows the above mentioned.

Micro-algal biomass is a versatile raw material that can potentially be used as a source for a range of non-fuel and fuel products, including bio-oils and proteins, high value chemicals and ingredients, food and feed, fertilizers and fuels and as a result of their growth, oxygen is released into the atmosphere as shown in Figure 4.

4. CONCLUSIONS AND DISCUSSION

While CCS (carbon capture and storage) removes carbon from the economy, CCU (carbon capture and utilization) turns waste CO₂ emissions into valuable products that can be sold. By utilizing CO₂ it is possible to retain carbon within a cycle. Many different products could be obtained from captured CO₂ in industries like food, products, plastic, extractant, chemical, refrigerants, as well as fire suppression, fuel recovery and biological conversion chemicals.

Captured CO₂ can be used as a commercial product, either directly or after conversion. Its direct utilization is not widely used, except for EOR, because it requires in most cases a high purity of this gas, especially in the food industry, horticulture and pharmaceutical industry. Captured CO₂ is commonly used after conversion and till now the most used conversions methods include chemical transformation into fuels, CO₂ absorption by microalgae to generate biomass and mineral carbonation.

CCS technologies and CCU technologies have developed as a necessity for combating climate change, mainly due to accelerated increase of CO₂ emissions. These technologies are temporary solutions. CCS has the advantage of safe storage in underground natural CO₂ environments over a certain
period of time and CCU has the economic advantage, being an activity that generates other profits. The main disadvantages of CCU technologies refer to the release of CO₂ in the atmosphere after the end of product lifetime (e.g. through burning of fuels) diminishing the real carbon abatement and to the large energy consumption associated with the conversion processes. In order to eliminate these disadvantages and to mitigate climate change, future research should aim to develop materials and products for which the raw material is CO₂ and such products should have longer life to enable the storage of CO₂ in a longer cycle (Cuéllar-Franca, Azapagic, 2015).

REFERENCES


Geo-Eco-Marina 24/2018